

## Biological membranes and ion channels

Reading:

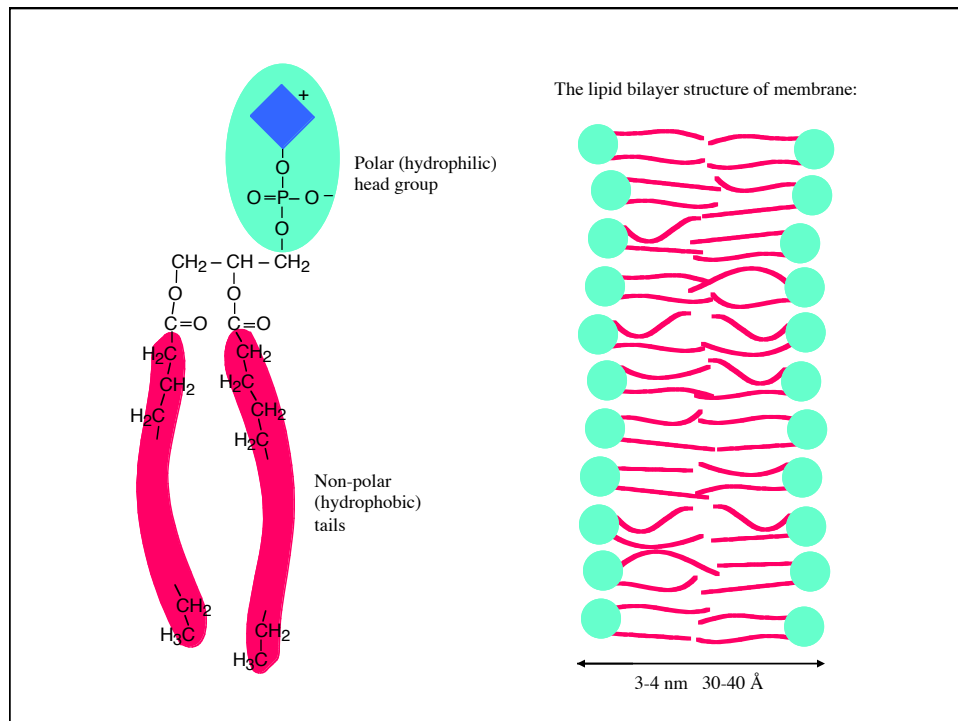
Hille (3rd ed.) chaps 10, 13, 17

Doyle et al. The Structure of the Potassium Channel: Molecular Basis of K<sup>+</sup> Conduction and Selectivity. *Science* 280:70-77 (1998).

Miyazawa et al. Structure and gating mechanism of the NACH receptor pore *Nature* 423:949-955 (2003).

Long et al. Voltage Sensor of Kv1.2: Structural Basis of Electromechanical Coupling *Science* 309:897-902 and 309:903-907 (2005).

Payandeh et al. The crystal structure of a voltage-gated sodium channel. *Nature* 475:353-359 (2011).



Membranes are stabilized and ions are unable to permeate membranes in significant numbers because of **hydrogen bonding** in aqueous solutions.

$$\begin{array}{c}
 \text{H} \\
 \diagup \quad \diagdown \\
 \text{O} \\
 \diagdown \quad \diagup \\
 \text{H}
 \end{array}$$

Water molecule dipole

H-bonded head region  
Tail region not H-bonded, energetically unfavorable!

Sodium ion stabilized in solution by H-bonds (176 RT bond energy/mole)

Same ion in membrane interior, no H-bonds. A high-energy state.

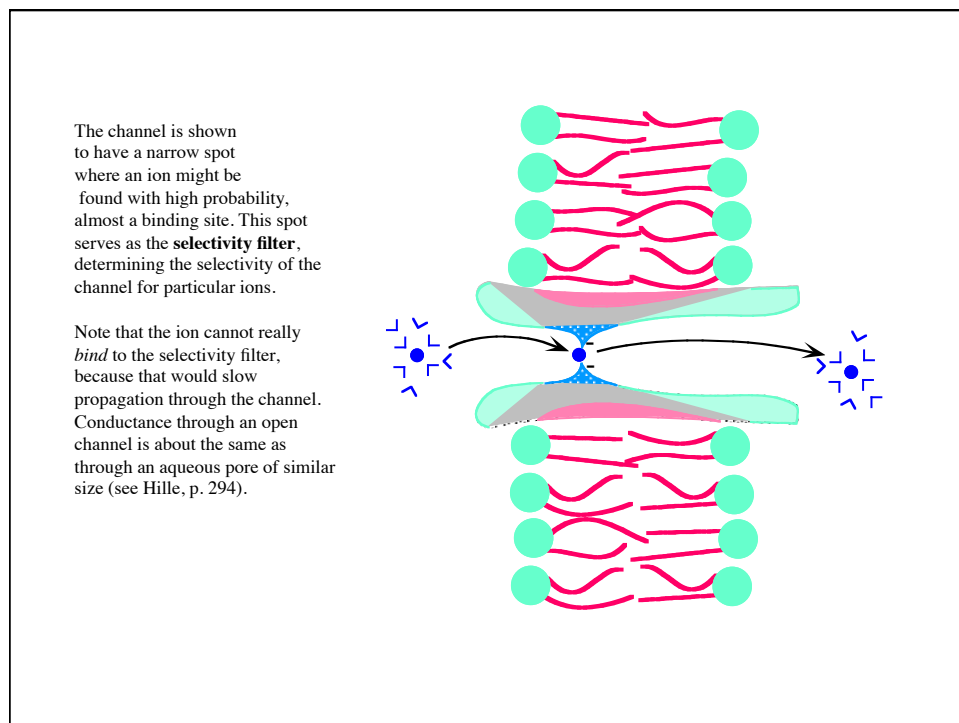
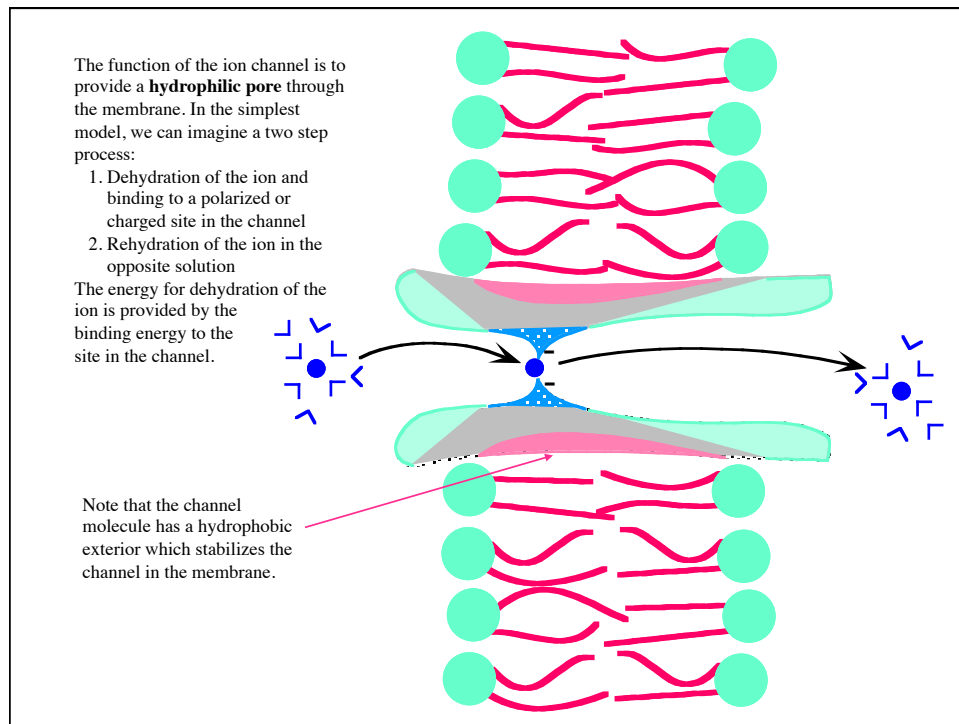
Pure lipid bilayers can be created artificially and have electrical characteristics like the circuit at right. The capacitance is about the same as for real nerve membrane, but real nerve membrane has a resistance several orders of magnitude smaller, about  $10^3 - 10^5$  ohm-cm<sup>2</sup>

1  $\mu\text{fd}/\text{cm}^2$   
inside ——— outside  
10<sup>8</sup>  $\Omega\text{-cm}^2$

The reason for the difference, of course, is that membrane contains **ion channels, transporters** and other proteins that provide specialized ionic conduction pathways through the membrane.

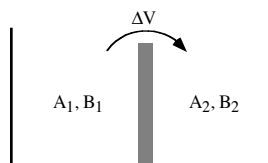
10 nm

Nicholls et al., 2001



**Selectivity** refers to the relative permeability of a membrane for different ions. Selectivity can be measured in various ways, but a simple (in principle) measurement is to use the GHK diffusion potential equation, applied to a situation in which only two permeable ions are present. Suppose the ions are both monovalent cations. Then in steady state,

$$\begin{aligned}\Delta V &= \frac{RT}{F} \ln \frac{P_A A_1 + P_B B_1}{P_A A_2 + P_B B_2} \\ &= \frac{RT}{F} \ln \frac{A_1 + \frac{P_B}{P_A} B_1}{A_2 + \frac{P_B}{P_A} B_2}\end{aligned}$$



The diffusion potential equation can be solved for the relative permeability  $P_B/P_A$  as below. Then, by measuring the membrane potential with various concentrations of ions, the relative permeability can be calculated.

$$\frac{P_B}{P_A} = \frac{B_1 - e^{F\Delta V/RT} B_2}{e^{F\Delta V/RT} A_2 - A_1}$$

Voltage gated ion channels show considerable selectivity, inferred from membrane potential experiments.

TABLE 14.2 Permeability Ratios,  $P_X/P_{Na}$ , for Na Channels<sup>a</sup>

Ion	Frog node <sup>1</sup>	Frog muscle <sup>2</sup>	Squid axon <sup>3</sup>	<i>Myxicola</i> axon <sup>4</sup>
H <sup>+</sup>	252 <sup>5</sup>	—	>2 <sup>6</sup>	—
Na <sup>+</sup>	1.0	1.0	1.0	1.0
Li <sup>+</sup>	0.93	0.96	1.1	0.94 <sup>9</sup>
Ca <sup>2+</sup>	<0.11	<0.093	0.1 <sup>8</sup>	0.1
K <sup>+</sup>	0.086	0.048	0.083	0.076 <sup>9</sup>
Rb <sup>+</sup>	<0.012	—	0.025	—
Cs <sup>+</sup>	<0.013	—	0.016	—
TMA	<0.005	<0.008	—	—

TABLE 14.4 Permeability Ratios,  $P_X/P_{Ca}$ , for L-Type Ca Channels<sup>a</sup>

Ion	$P_X/P_{Ca}$	Ion	$P_X/P_{Ca}$
Ca <sup>2+</sup>	1.0	Li <sup>+</sup>	1/424
Sr <sup>2+</sup>	0.67	Na <sup>+</sup>	1/1170
Ba <sup>2+</sup>	0.40	K <sup>+</sup>	1/3000
		Cs <sup>+</sup>	1/4200

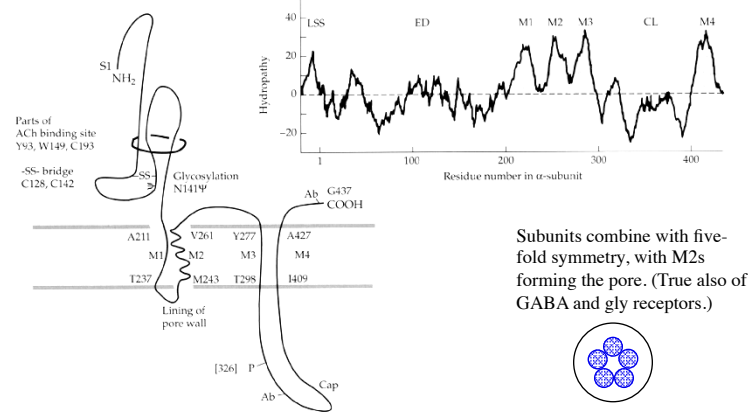
TABLE 14.3 Permeability Ratios,  $P_X/P_K$ , for Several Types of K Channels<sup>a</sup>

Ion	Delayed rectifier			Inward rectifier	M current	BK K(Ca)	SK K(Ca)
	Frog node <sup>1</sup>	Frog muscle <sup>2</sup>	Snail neuron <sup>3</sup>	Starfish egg <sup>4</sup>	Frog neuron <sup>5</sup>	Rat muscle <sup>6</sup>	Rat chromaffin <sup>7</sup>
K <sup>+</sup>	1.0	1.0	1.0	1.0	1.0	1.0	1.0
Rb <sup>+</sup>	0.91	0.95	0.74	0.35	0.94	0.67	0.81
Cs <sup>+</sup>	<0.077	<0.11	0.18	<0.03	0.10	<0.05	0.16
Li <sup>+</sup>	<0.018	<0.02	0.09	—	<0.004	<0.02	<0.005
Na <sup>+</sup>	<0.010	<0.03	0.07	<0.03	<0.004	<0.01	<0.005

Tables from Hille, 2001

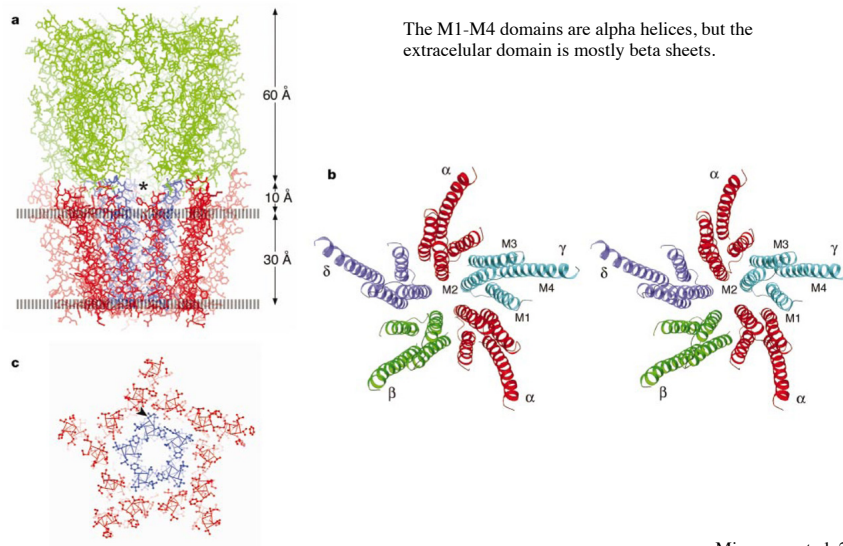
Ion channels are protein molecules with extracellular and intracellular domains and transmembrane domains. The **nicotinic acetylcholine receptor channel** subunit is diagrammed below. Note the **transmembrane segments**, denoted M1 - M4.

- Evidence for channel structure:  
 1. Hydropathy plots  
 2. Binding sites for various ligands



Hille, 2001

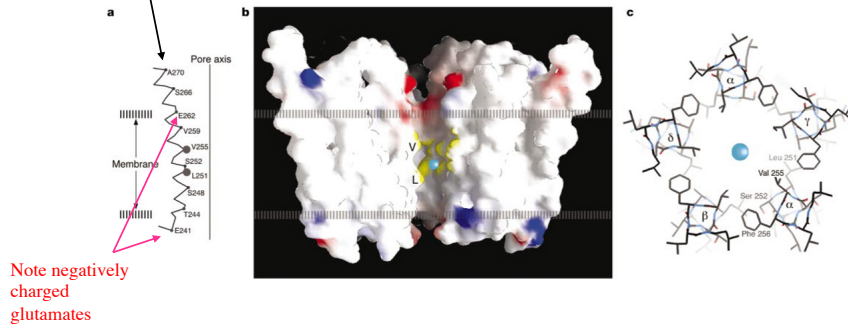
Using electron diffraction the structure of the NACHR was determined at 4 Å resolution. The M2 segments form a loose pore (blue) with substantial aqueous space in the membrane (between blue and red). The large extracellular domain contains the ACh binding site (green).



Miyazawa et al. 2003

A space-filling model of the transmembrane part of the channel shows the pore, which is large. Red regions are negatively charged or polarized, to attract ions to the channel. The yellow region around the valine and leucine at positions 255 and 251 is non-polar and is the gate, which closes or opens in response to ACh binding.

The amino acids making up the pore.



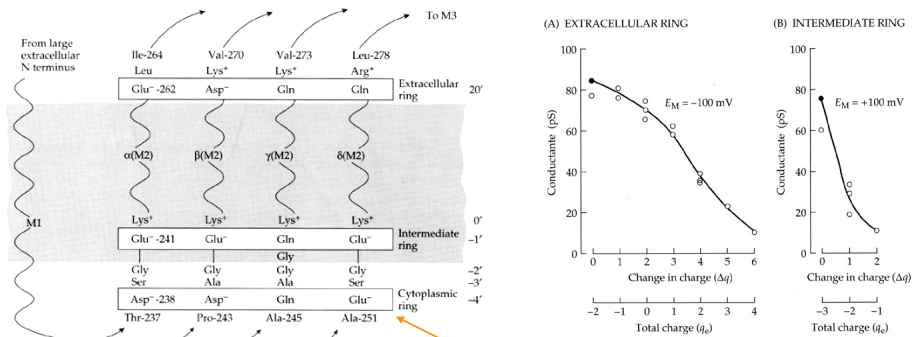
Miyazawa et al. 2003

The channel of the NACHR is formed by M2 transmembrane segments.

Three rings of negative charge in the pore control its permeation characteristics.

1. Extracellular and cytoplasmic rings contain 2 and 3 charges; they appear to attract cations to the channel mouth.
2. An intermediate ring near the narrow spot in the pore contains 3 charges. It is more important in determining channel conductance than the external rings.

(Note the rings of positively charged residues; presumably these are rotated on the M2 a helix out of the channel pore.)

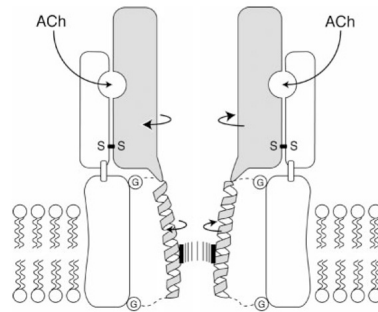


Originally thought to be in the pore, but probably not.

(Hille, 2001)

Model for gating of the NACHR. Binding of ACh causes the gray part of the extracellular domain to rotate as drawn, producing a rotation and realignment of the M2 segments increasing or decreasing the size of the pore in the vicinity of V255 and L251.

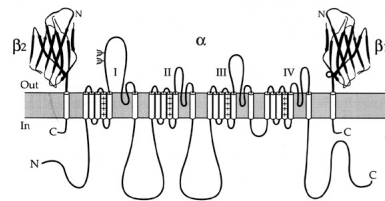
2010, lect #5



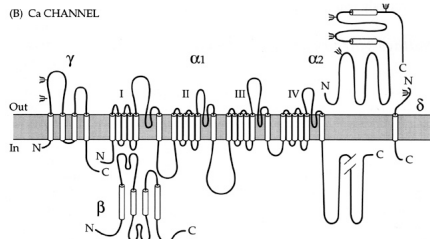
Miyazawa et al. 2003

**Voltage-gated cation channels** consist of four subunits, each of which has 6 transmembrane segments and a **pore loop**. In sodium and calcium channels, the four subunits are part of the same molecule. In potassium channels, they are different molecules. Also shown are  $\beta$  and  $\beta_1$  subunits, separate molecules that bind to the channel and change its properties.

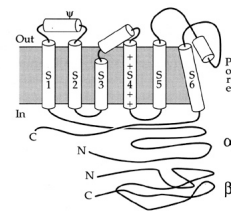
(A) Na CHANNEL



(B) Ca CHANNEL



(C) ONE QUARTER OF A K CHANNEL

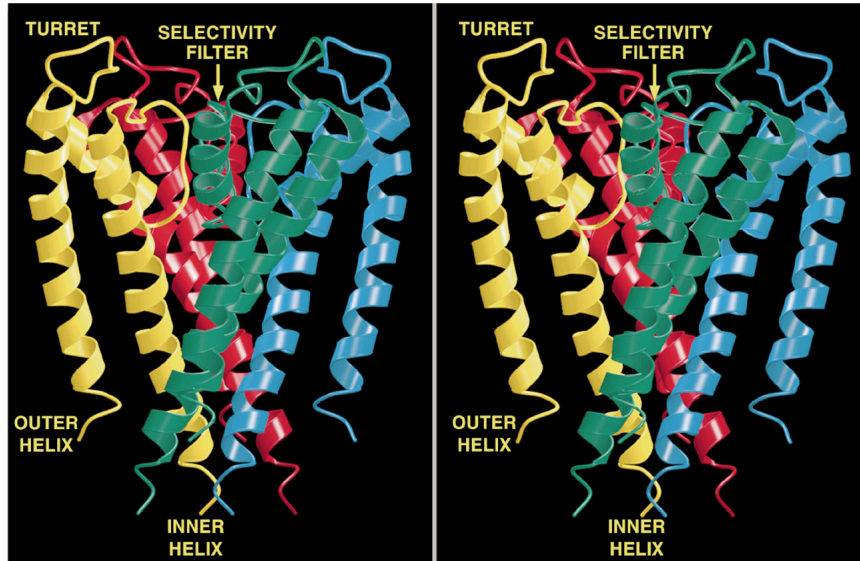


The resulting channel has four-fold symmetry



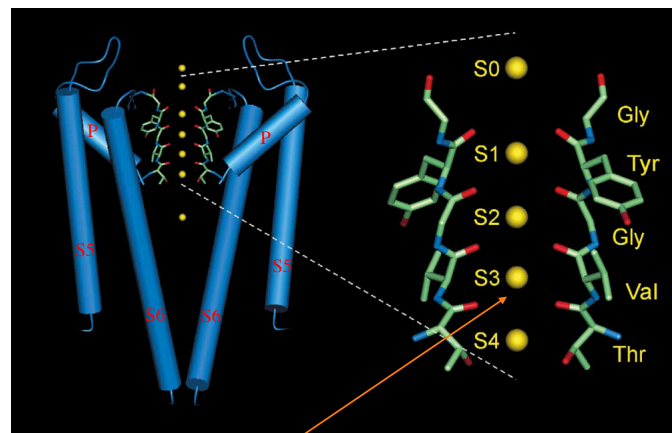
(Hille, 2001)

The structure of potassium channels has been inferred from the KcsA channel, a bacterial K channel having only the S5, S6, and P domains. The pore is formed by the S6 and P domains.



(Doyle et al, *Science* 280:69, 1998)

The **KcsA selectivity filter** consists of five amino acids (TVGYG) on the S6-P connector. The potassium ions (yellow balls) interact with the peptide backbone carbonyl oxygens (small red balls), not the side chains. There are four stable positions for K<sup>+</sup> ions (S1-S4), plus a fifth, entry, position. Only two are occupied at a time, because of electrostatic interaction among the K<sup>+</sup>s. **Selectivity** is determined by the properties of K<sup>+</sup> ion interaction with this structure.



Note that the ions are in **single file**. This fact is consistent with a number of previous biophysical observations

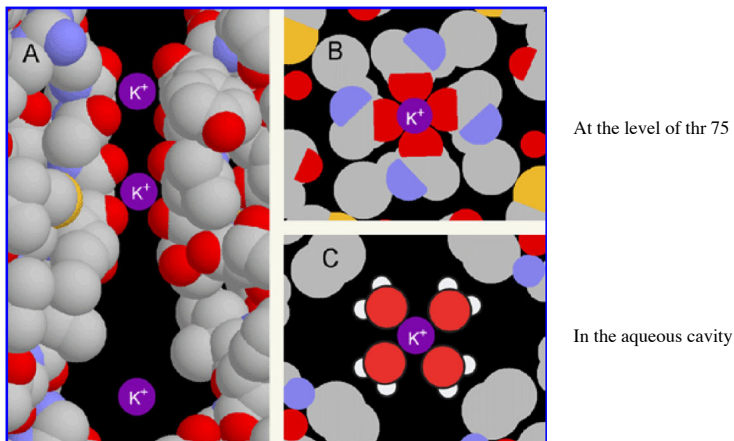
Sansom et al. 2002)

### How might the KcsA selectivity filter work?

Potassium ions (dehydrated) are stabilized in the pore region (A) by negatively charged carbonyl groups from the protein making up the wall of the selectivity filter.

Presumably, the  $K^+$  ions "just fit" into the cross section of the pore (B). The electrostatic binding between the  $K^+$  and the carbonyls replaces the H-bonding in the aqueous environment, facilitating entry of  $K^+$  into the channel.

In the cavity of the KcsA channel, there is room for the potassium ions to carry a hydration shell, facilitating transport of ions in and out of the channel on the cytoplasmic side.

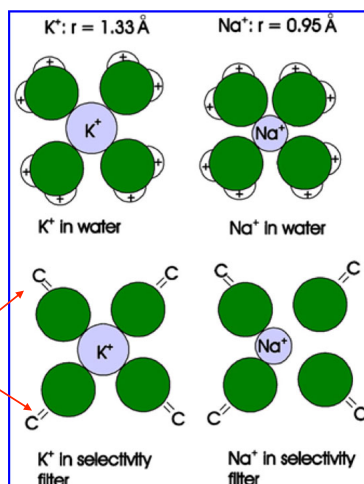


Armstrong, 2003

By contrast,  $Na^+$  ions, which have a smaller ionic radius, do not bind efficiently to all four carbonyls, as shown in the schematic cross sections at right.

Because the binding energy varies inversely with the distance between charges, Na is less stabilized in the selectivity filter than K, and is less likely to escape from an aqueous hydration shell into the pore.

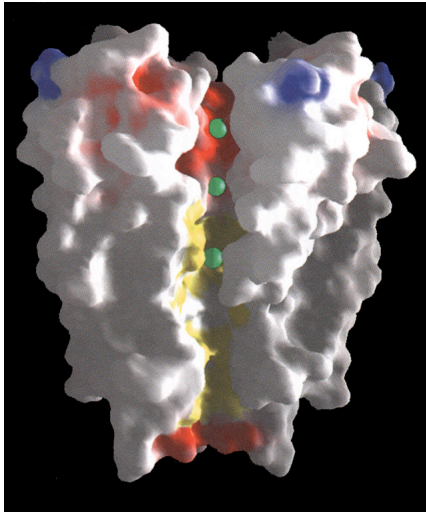
Interactions with surrounding moieties stiffen the selectivity filter so it can't collapse on the  $Na^+$  ion



Armstrong, 2003

(Doyle et al, *Science* 280:69, 1998)

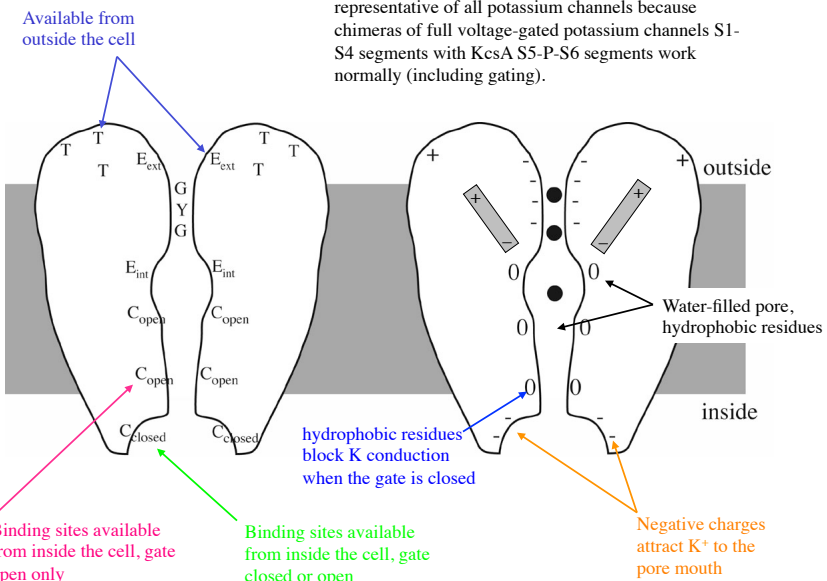
At right is a space-filling model of the KcsA channel, showing the pore. Ions (green balls) tend to occupy three sites in the channel, two in the **selectivity filter** and one in a **pool of water** in the center of the channel. Note the negative charges (red) at the two ends of the pore, which attract ions to the channel's entrance. Subsequently, it has been shown that the hydrophobic (yellow) narrow spot on the cytoplasmic side of the channel is the **gate**.



Sansom et al. (2002)

red - charge; blue + charge; yellow hydrophobic

The structure of the KcsA channel is believed to be representative of all potassium channels because chimeras of full voltage-gated potassium channels S1-S4 segments with KcsA S5-P-S6 segments work normally (including gating).



Available from outside the cell

Binding sites available from inside the cell, gate open only

Binding sites available from inside the cell, gate closed or open

hydrophobic residues block K conduction when the gate is closed

Water-filled pore, hydrophobic residues

Negative charges attract  $K^+$  to the pore mouth

outside

inside

