## **Aura and Stratospheric Ozone**

Richard S. Stolarski
Johns Hopkins University
Baltimore, MD

(Emeritus at GSFC)





## **Stratospheric Ozone**

Ozone is a tri-atomic form of oxygen, O<sub>3</sub>



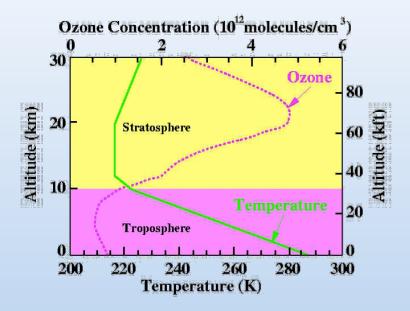
- Ozone is produced by ultraviolet radiation from the sun
- Ozone is destroyed by chemical reactions that recombine ozone to molecular oxygen

$$(0 + 0_3 \rightarrow 0_2 + 0_2)$$

 Ozone is a "renewable resource": its concentration is determined by a balance of production and loss

## **Stratospheric Ozone**

Ozone forms a layer in the stratosphere



Ozone layer shields the surface from UV radiation

### The Fluorocarbon-Ozone Problem

 Ozone destruction can be enhanced by catalytic reactions of the oxides of chlorine, bromine, nitrogen, and hydrogen

 Most chlorine compounds are soluble and do not reach the stratosphere

### The Fluorocarbon-Ozone Problem

 Chlorofluorocarbons (CFCs) are industrially-produced chemicals that were used in refrigeration, aerosol sprays, foam-blowing, etc

$$CFC-11 = CFCl3$$

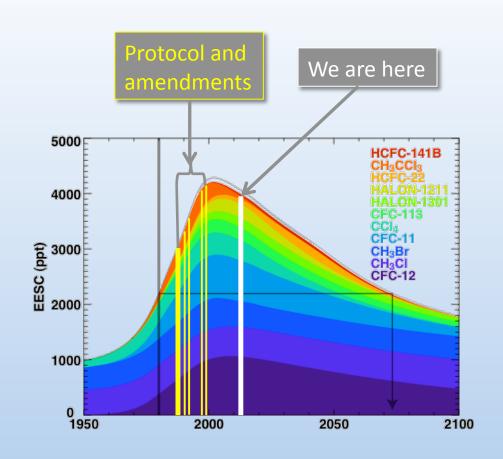
$$CFC-12 = CF2Cl2$$

$$CFC-113 = C2F3Cl3$$

 They are insoluble, unreactive, and non-absorbing of visible radiation; they can reach the stratosphere where UV radiation releases the chlorine

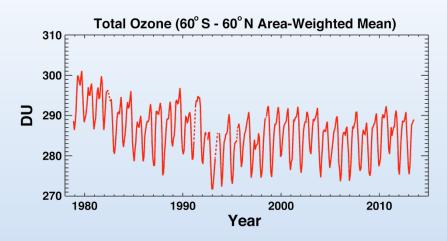
### The Fluorocarbon-Ozone Problem

- Fluorocarbons emissions were growing rapidly in the 1970s and 1980s
- The Montreal Protocol
   (1987) and its amendments
   (1990, 1992, 1997, 1999)
   began limiting production
- Chlorine levels are declining



## How does Aura enter this picture?

- Ozone concentrations were decreasing
- They appear to have leveled off
- Was this because of the Montreal Protocol?
- What will happen in the future?



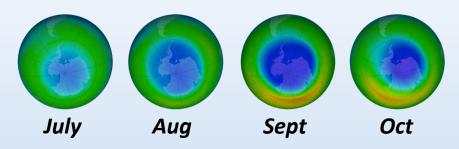
Data from the NASA/NOAA SBUV series of instruments

Can we explain the general shape of this curve? Can we explain the deviations from this shape?

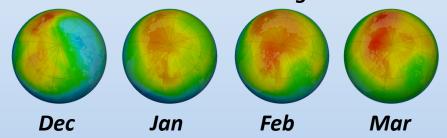
## How does Aura enter this picture?

- "Ozone hole" observed in southern polar spring
- Very different behavior in north polar region
- What causes difference?
- What causes year-toyear variation, particularly in NH?

#### Southern Polar Region

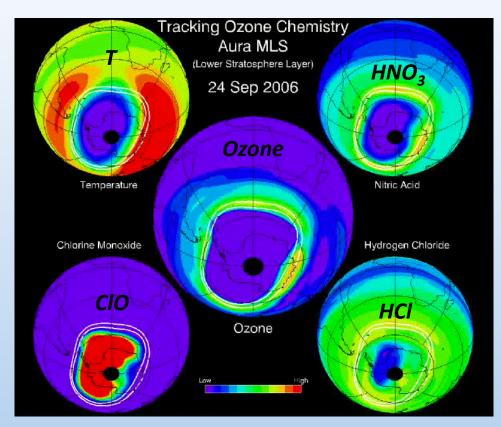


#### Northern Polar Region



Graphics from ozonewatch.gsfc.nasa.gov

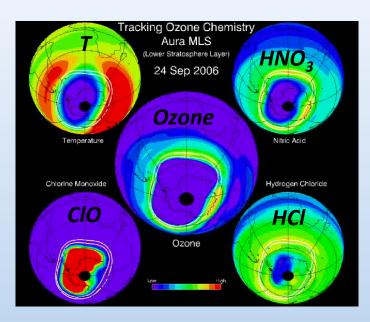
## Aura MLS Measures Ozone and the Chemicals that Determine its Concentration



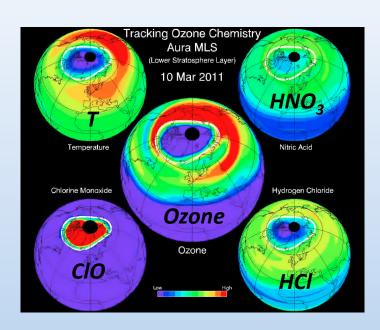
South Polar 24 September 2006

- Temperature: cold T results in PSCs
- Nitric acid: HNO<sub>3</sub> removed by PSCs
- HCl: reservoir for chlorine
- CIO: active form of chlorine for catalysis
- Ozone: reduced by chlorine catalysis

# We can contrast the northern and southern polar regions

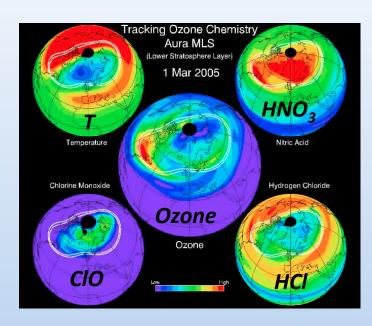


South Polar 24 September 2006

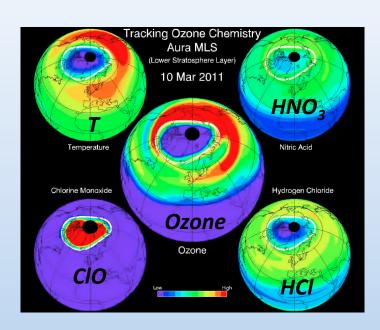


North Polar 10 March 2011

# We can observe radically different northern polar winters

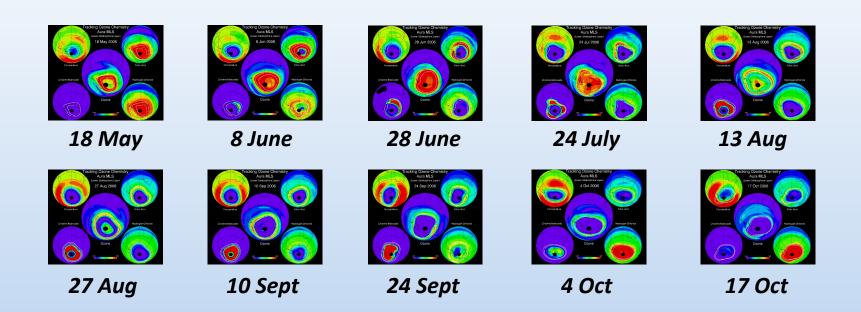


North Polar 1 March 2005

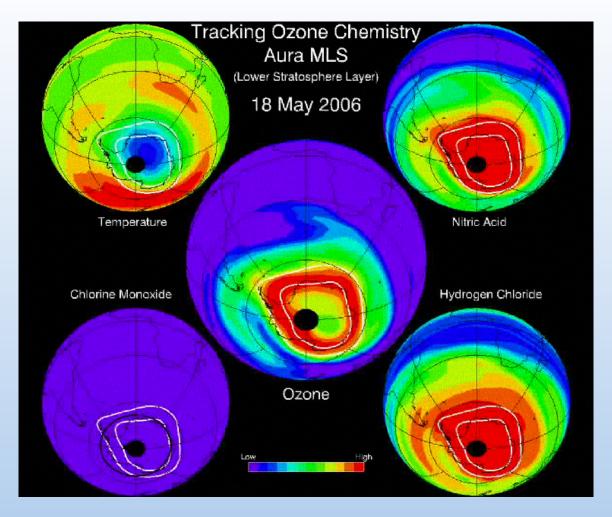


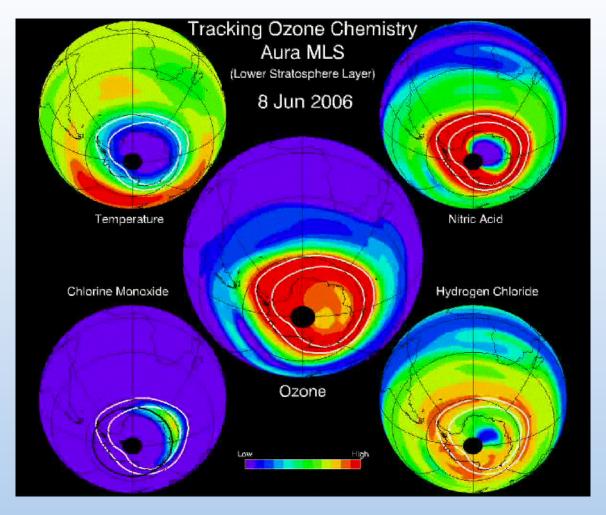
North Polar 10 March 2011

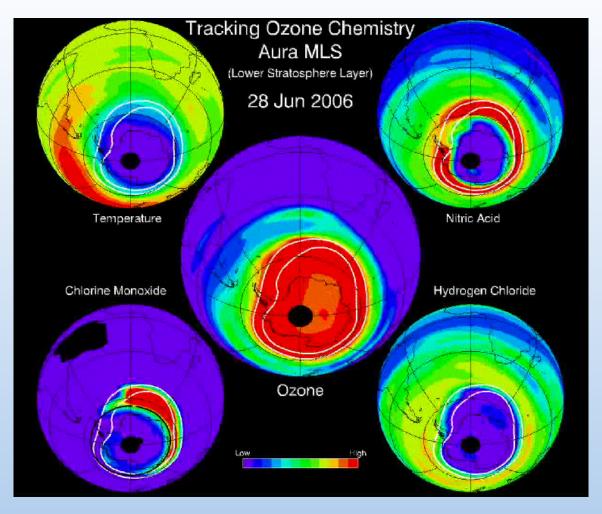
# We can observe a daily sequence over an entire polar winter

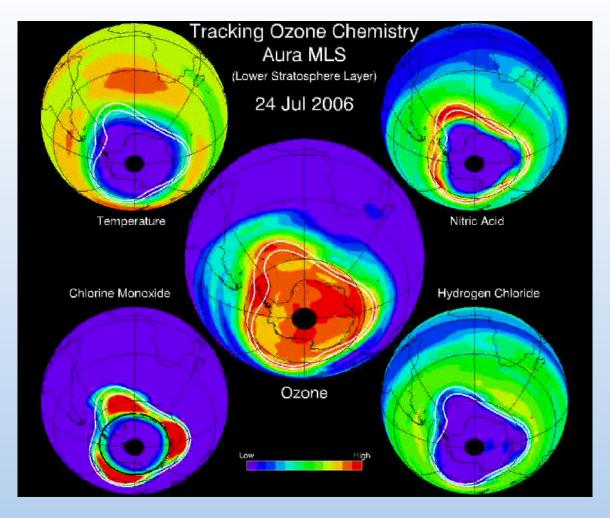


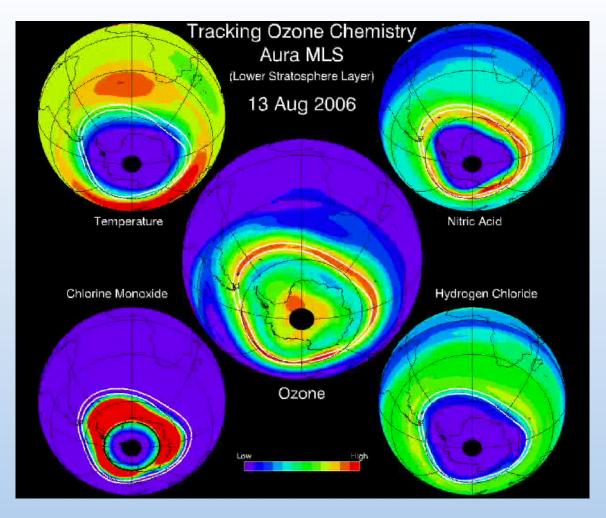
Next 10 slides will show this sequence for the south polar region

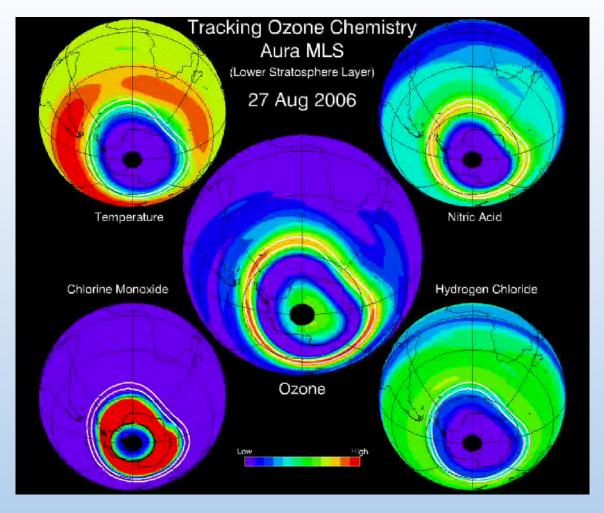


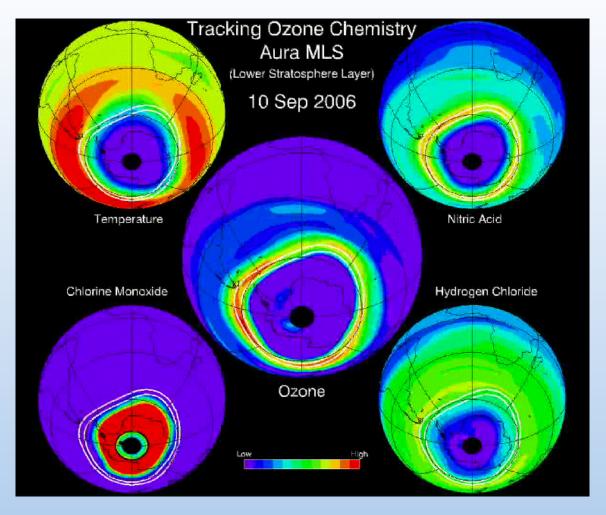


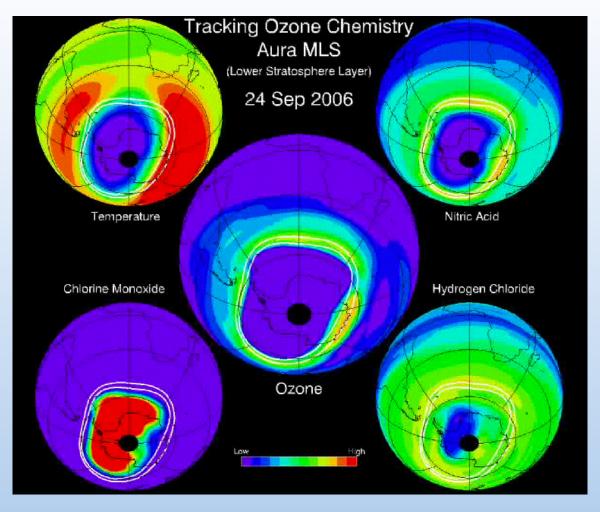


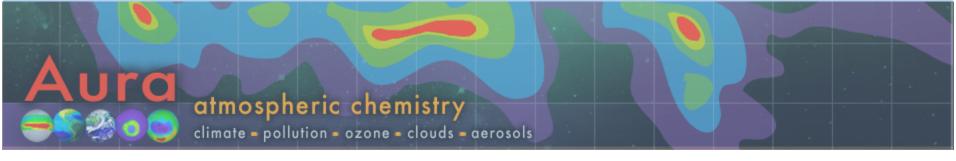


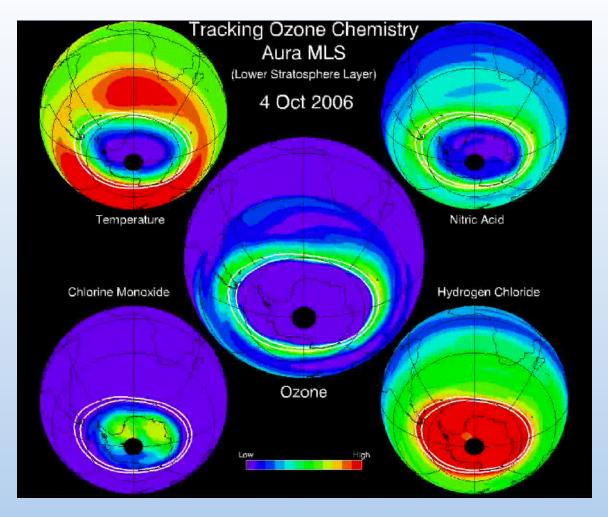




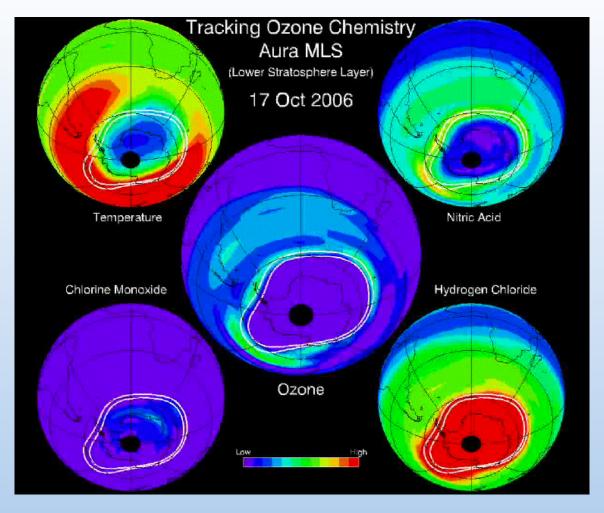












## **Attribution and future prediction**

- With Aura we have been able to observe the interaction of chemistry with the motions of the stratosphere over the entire globe under a wide variety of conditions
- We can use models, together with Aura data to attribute variations in ozone to cause
- We thus have the tools to make future predictions of ozone with a greater degree of confidence due to the measurements on Aura
- If future ozone deviates from expectation, we have the capability to diagnose the cause